



PROSPECTS: Prebiotic mOlecules from SPacE to ComeTS

Vianney Taquet

Osservatorio Astrofisico di Arcetri Marie Skłodowska-Curie ASTROFIT2 Fellow

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The formation of low-mass stars



How to observe molecules in star-forming regions ?



How to observe molecules in star-forming regions ?

2. Millimetric spectroscopy in emission to study the molecular gas



How to observe molecules in star-forming regions ?



Expertises in Astrochemistry



Main projects during ASTROFIT2 program



Bright cold methanol emission in Barnard 5 Taquet, Wirström, Charnley et al. (In prep.)

Methanol: a key molecule for organic chemistry

High methanol abundances both in interstellar ices and gas



High-mass protostar AFGL7009S

Dartois et al. (1998), Boogert et al. (2015)

Low-mass protostar IRAS4A observed with NOEMA



De Simone et al., Taquet et al. (in prep.)

Methanol: a key molecule for organic chemistry

Methanol is likely the mother molecule of many interstellar Complex Organic Molecules



Molecular mapping of the Barnard 5 molecular cloud





Bright unexpected methanol emission in Barnard 5

Cold methanol is detected in a "secondary" part of the cloud with relatively high abundances



Molecular mapping of the Barnard 5 molecular cloud



Multi-variate analysis of the molecular emission

Decomposition of initial dataset into principal components: orthogonal variables that maximise the observed variance



Methanol emission triggered by cloud-cloud emission ?

Methanol emission is detected where two H₂CO components collide:



Sulphur chemistry along molecular outflows

Taquet, Codella, De Simone, López-Sepulcre, Pineda et al. (to be subm.)

Why sulfur chemistry ?

Potential chemical clock around protostars because abundances of "bright" species (i.e. CS, SO, H₂S, OCS, SO₂) are thought to evolve strongly with time



Charnley (1997)

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Can we use sulfur chemistry to assess the physical and chemical conditions of molecular outflows ?



Charnley (1997)

The NGCC1333-IRAS4A protostellar system



Dissecting the outflows with clustering methods

Decomposition of the outflows based on their molecular emission only:

- K+Means algorithm

26"

- 9 features (mom. 0, 1, and 2 of OCS, SO, and SiO)

3

0

- Spatial information not used[®]
- Number of *clusters K* as input

3h29m10.5s

R.A. [J2000]

Jec. [J2000]



Chemical interpretation



Methanol deuteration in low-mass protostars

Taquet, Bianchi, Codella, Persson, Ceccarelli et al. (A&A, in press)

Gas phase [D]/[H] in dark clouds

Methanol and water are formed through hydrogenation reactions on cold interstellar grains

 \rightarrow their deuteration highly depends on atomic [D]/[H] in the gas

Key reaction for deuterium chemistry: $H_3^+ + HD \rightleftharpoons H_2D^+ + H_2 + \Delta E = 232 \text{ K}$

Deuteration in the gas (and in ices) is governed by:

- Temperature
- CO abundance
- Ortho/para ratio of H_2
- Moment of formation

Deuteration: powerful chemical tracer !

See Pagani et al. (1992); Roberts et al. (2003, 2004); Flower et al. (2006); Hugo et al. (2009), ...

Warm deuterated methanol towards low-mass protostars

Analysis of NOEMA and ALMA observations towards four low-mass protostars in different molecular clouds: Perseus, Orion, Ophiuchus



Methanol deuteration observed towards hot cores mostly regulated by the temperature of the progenitor cloud ?



SgrB2), Bøgelund+ (2018, NGC6334), Jørgensen+ (2018, IRAS16293)

Methanol deuteration observed towards hot cores mostly regulated by the temperature of the progenitor cloud ?

+ Other processes (time, warm ice or gas phase chemistry) at work ?



Predictions with the Taquet+ (2014) model but with constant physical conditions

MOMICE: MOdel for Multiphase Ice ChEmistry

Fortran90 code accompanied by Python scripts to run and analyse the simulations:



Three networks:

- Network 1: H₂O, CO₂, and CH₃OH formation without gas phase chemistry
- Network 2: ice deuteration
- Network 3: extended surface + KIDA gas phase networks

Four options:

- 1) Invididual simulations with constant physical conditions
- 2) Spatial evolution with evolving physical conditions
- 3) Model grid to explore the impact of physical conditions on ice chemistry
- 4) Sensitivity analysis to evaluate the impact of surface and chemical parameter uncertainties on ice chemistry

"Astronomical" and "experimental" versions

Processes included in the model

MOMICE multi-phase (bulk, surface, gas) astrochemical model Taquet et al. (2012, 2013, 2014, 2016); Dulieu et al. (2018)

1) Gas phase chemical network based on KIDA database



Constraining the models with experiments

Chemical modelling of laboratory experiments:

 \rightarrow validation of the formalism and constraints on surface/chemical parameters

Example for CO hydrogenation (Watanabe et al. 2004 and Fuchs et al. 2009) $E_d/E_b = 0.45 \pm 0.05$ $E_a(CO+H->HCO) = 4000 \pm 500$ K

- $E_a(H_2CO+H_2) = 4000 \pm 250 \text{ K}$
- $E_a(H_2CO+H->CH_3O) = 4750 \pm 250 \text{ K}$



Experiments from Watanabe et al. (2004)

Impact of physical conditions on ice composition



Impact of uncertainties on ice composition

10% uncertainty on binding energies, diffusion-to-binding energy ratio, and activation energies

 $n_{\rm H} = 2 \times 10^4 \, {\rm cm}^{-3}, \, {\rm T} = 10 \, {\rm K}$



See recent works by Penteado et al. (2017), Holdship et al. (2018)

Impact of uncertainties on ice composition

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Main activities during the PROSPECTS project

- ~ 25 publications:

- 4 first-author publications
- 11 others with significant contributions

- Oral contributions:

- 4 seminars
- 5 invited talks

- Visits and collaborations:

- Two visits at Paris Observatory (F. Dulieu)
- Many visits at University of Grenoble (C. Ceccarelli)
- Collaborations with Leiden Observatory, NASA Goddard